transmission and it has a minimum torque even when all the driven hydraulic actuators and motors are stationary and the engine is at idle, then you may use this declared minimum torque as a reference torque value instead of any reference torque value generated under paragraph (d)(1) or (2) of this section that is between zero and this declared minimum torque.

(e) Generating reference power values from normalized duty cycle powers. Transform normalized power values to reference speed and power values using your map of maximum power versus speed.

(1) First transform normalized speed values into reference speed values. For a given speed point, multiply the corresponding % power by the mapped power at maximum test speed,  $f_{\text{ntest}}$ , unless specified otherwise by the standard-setting part. The result is the reference power for each speed point,  $P_{ref}$ . Convert these reference powers to corresponding torques for operator demand and dynamometer control and for duty cycle validation per 1065.514. Use the reference speed associated with each reference power point for this conversion. As with cycles specified with % torque, linearly interpolate between these reference torque values generated from cycles with % power.

(2) Permissible deviations for any engine. If your engine does not operate

below a certain power under normal inuse conditions, you may use a declared minimum power as the reference value instead of any value denormalized to be less than the declared value. For example, if your engine is directly connected to a propeller, it may have a minimum power called idle power. In this case, you may use this declared minimum power as a reference power value instead of any reference power value generated per paragraph (e)(1) of this section that is from zero to this declared minimum power.

[73 FR 37324, June 30, 2008, as amended at 73 FR 59330, Oct. 8, 2008; 75 FR 23045, Apr. 30, 2010; 76 FR 57453, Sept. 15, 2011; 78 FR 36398, June 17, 2013; 79 FR 23783, Apr. 28, 2014]

# § 1065.630 Local acceleration of gravity.

(a) The acceleration of Earth's gravity,  $a_{\rm g}$ , varies depending on the test location. Determine  $a_{\rm g}$  at your location by entering latitude, longitude, and elevation data into the U.S. National Oceanographic and Atmospheric Administration's surface gravity prediction Web site at http://www.ngs.noaa.gov/cgi-bin/grav\_pdx.prl.

(b) If the Web site specified in paragraph (a) of this section is unavailable, you may calculate  $a_{\rm g}$  for your latitude as follows:

$$a_{g} = 9.7803267715 \cdot [1 + 5.2790414 \cdot 10^{-3} \cdot \sin^{2}(\theta) + 2.32718 \cdot 10^{-5} \cdot \sin^{4}(\theta) + 1.262 \cdot 10^{-7} \cdot \sin^{6}(\theta) + 7 \cdot 10^{-10} \cdot \sin^{8}(\theta)]$$

Eq. 1065.630-1

Where:

 $\theta = \text{Degrees}$  north or south latitude.

Example:

 $\theta=45^{\circ}$ 

[79 FR 23784, Apr. 28, 2014]

#### § 1065.640 Flow meter calibration calculations.

This section describes the calculations for calibrating various flow meters. After you calibrate a flow meter using these calculations, use the calculations described in §1065.642 to calculate flow during an emission test. Paragraph (a) of this section first describes how to convert reference flow

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meter outputs for use in the calibration equations, which are presented on a molar basis. The remaining paragraphs describe the calibration calculations that are specific to certain types of flow meters.

(a) Reference meter conversions. The calibration equations in this section use molar flow rate,  $\dot{n}_{\rm ref}$ , as a reference quantity. If your reference meter outputs a flow rate in a different quantity, such as standard volume rate,  $\dot{V}_{\text{stdref}}$ , ac-

tual volume rate,  $\dot{V}_{
m actref}$ , or mass rate,  $\dot{m}_{\rm ref}$ , convert your reference meter output to a molar flow rate using the following equations, noting that while values for volume rate, mass rate, pressure, temperature, and molar mass may change during an emission test, you should ensure that they are as constant as practical for each individual set point during a flow meter calibration:

$$\dot{n}_{\rm ref} = \frac{\dot{V}_{\rm stdref} \cdot p_{\rm std}}{T_{\rm std} \cdot R} = \frac{\dot{V}_{\rm actref} \cdot p_{\rm act}}{T_{\rm act} \cdot R} = \frac{\dot{m}_{\rm ref}}{M_{\rm mix}}$$

Eq. 1065.640-1

Where:

 $\dot{n}_{\mathrm{ref}}$  = reference molar flow rate.  $\dot{V}_{\mathrm{stdref}}$  = reference volume flow rate, corrected to a standard pressure and a standard temperature.

 $\dot{V}_{\rm actref}$  = reference volume flow rate at the actual pressure and temperature of the flow rate.

 $\dot{m}_{\rm ref}$  = reference mass flow.

 $p_{\rm std}$  = standard pressure.

 $p_{\rm act}$  = actual pressure of the flow rate.

 $T_{\rm std} = {\rm standard}$  temperature.

 $T_{\rm act} = {
m actual}$  temperature of the flow rate.

R = molar gas constant.

 $M_{\text{mix}}$  = molar mass of the flow rate.

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Example 1:

$$\dot{V}_{\text{stdref}} = 1000.00 \text{ ft}^3/\text{min} = 0.471948 \text{ m}^3/\text{s}$$

$$p_{\rm std}$$
 = 29.9213 in Hg @ 32 °F = 101.325 kPa = 101325 Pa = 101325 kg/(m·s<sup>2</sup>)

$$T_{\rm std} = 68.0 \, {}^{\circ}\text{F} = 293.15 \, \, \text{K}$$

$$R = 8.314472 \text{ J/(mol \cdot K)} = 8.314472 \text{ (m}^2 \cdot \text{kg)/(s}^2 \cdot \text{mol \cdot K)}$$

$$\dot{n}_{\text{ref}} = \frac{0.471948 \cdot 101325}{293.15 \cdot 8.314472}$$

$$\dot{n}_{\rm ref} = 19.619 \text{ mol/s}$$

Example 2:

$$\dot{m}_{\rm ref} = 17.2683 \text{ kg/min} = 287.805 \text{ g/s}$$

$$M_{\rm mix} = 28.7805 \text{ g/mol}$$

$$\dot{n}_{\rm ref} = \frac{287.805}{28.7805}$$

$$\dot{n}_{\rm ref} = 10.0000 \text{ mol/s}$$

- (b) <u>PDP calibration calculations</u>. For each restrictor position, calculate the following values from the mean values determined in § 1065.340, as follows:
- (1) PDP volume pumped per revolution,  $V_{rev}$  (m<sup>3</sup>/r):

$$V_{\text{rev}} = \frac{\overline{\dot{n}}_{\text{ref}} \cdot R \cdot \overline{T}_{\text{in}}}{\overline{p}_{\text{in}} \cdot \overline{f}_{\text{nPDP}}}$$

Eq. 1065.640-2

Example:

$$\overline{\dot{n}}_{\rm ref} = 25.096 \text{ mol/s}$$

$$R = 8.314472 \text{ J/(mol \cdot K)} = 8.314472 \text{ (m}^2 \cdot \text{kg)/(s}^2 \cdot \text{mol \cdot K)}$$

$$\overline{T}_{in} = 299.5 \text{ K}$$

$$\overline{p}_{in}$$
 = 98.290 kPa = 98290 Pa = 98290 kg/(m·s<sup>2</sup>)

$$\overline{f}_{\text{nPDP}} = 1205.1 \text{ r/min} = 20.085 \text{ r/s}$$

$$V_{rev} = \frac{25.096 \cdot 8.314472 \cdot 299.5}{98290 \cdot 20.085}$$

$$V_{rev} = 0.03166 \text{ m}^3/\text{r}$$

(2) PDP slip correction factor,  $K_s$  (s/r):

$$K_{\rm s} = \frac{1}{\overline{f}_{\rm nPDP}} \cdot \sqrt{\frac{\overline{p}_{\rm out} - \overline{p}_{\rm in}}{\overline{p}_{\rm out}}}$$

Eq. 1065.640-3

Example:

$$\overline{f}_{nPDP} = 1205.1 \text{ r/min} = 20.085 \text{ r/s}$$

$$\overline{p}_{\text{out}} = 100.103 \text{ kPa}$$

$$\bar{p}_{in} = 98.290 \text{ kPa}$$

$$K_{\rm s} = \frac{1}{20.085} \cdot \sqrt{\frac{100.103 - 98.290}{100.103}}$$

$$K_{\rm s} = 0.006700 \text{ s/r}$$

(3) Perform a least-squares regression of PDP volume pumped per revolution,  $V_{\rm rev}$ , versus PDP slip correction factor,

 $K_s$ , by calculating slope,  $a_1$ , and intercept,  $a_0$ , as described in § 1065.602.

(4) Repeat the procedure in paragraphs (b)(1) through (3) of this section for every speed that you run your PDP.

(5) The following example illustrates these calculations:

TABLE 1 OF § 1065.640—EXAMPLE OF PDP

| $ar{f}_{	ext{nPDP}}$ (revolution/s) | a <sub>1</sub><br>(m <sup>3</sup> /s) | a <sub>0</sub><br>(m <sup>3</sup> /revo-<br>lution) |
|-------------------------------------|---------------------------------------|---|
| 12.6                                | 0.841<br>0.831<br>0.809<br>0.788      | 0.056<br>-0.013<br>0.028<br>-0.061                  |

(6) For each speed at which you operate the PDP, use the corresponding slope,  $a_1$ , and intercept,  $a_0$ , to calculate flow rate during emission testing as described in §1065.642.

(c) Venturi governing equations and permissible assumptions. This section describes the governing equations and permissible assumptions for calibrating a venturi and calculating flow using a venturi. Because a subsonic venturi (SSV) and a critical-flow venturi (CFV) both operate similarly, their governing equations are nearly the same, except for the equation describing their pressure ratio, r (i.e., r<sub>SSV</sub> versus r<sub>CFV</sub>). These governing equations assume one-dimensional isentropic

inviscid compressible flow of an ideal gas. In paragraph (c)(4) of this section, we describe other assumptions that you may make, depending upon how you conduct your emission tests. If we do not allow you to assume that the measured flow is an ideal gas, the governing equations include a first-order correction for the behavior of a real gas; namely, the compressibility factor, Z. If good engineering judgment dictates using a value other than Z=1, you may either use an appropriate equation of state to determine values of Z as a function of measured pressures and temperatures, or you may develop your own calibration equations based on good engineering judgment. Note that the equation for the flow coefficient,  $C_{\rm f}$ , is based on the ideal gas assumption that the isentropic exponent,  $\gamma$ , is equal to the ratio of specific heats,  $C_p/C_v$ . If good engineering judgment dictates using a real gas isentropic exponent, you may either use an appropriate equation of state to determine values of  $\gamma$  as a function of measured pressures and temperatures, or you may develop your own calibration equations based on good engineering judgment. Calculate molar flow rate, as follows:

$$\dot{n} = C_{\rm d} \cdot C_{\rm f} \cdot \frac{A_{\rm t} \cdot p_{\rm in}}{\sqrt{Z \cdot M_{\rm mix} \cdot R \cdot T_{\rm in}}}$$

Eq. 1065.640-4

Where:

 $C_{\rm d}$  = discharge coefficient, as determined in paragraph (c)(1) of this section.

 $C_{\rm f}$  = flow coefficient, as determined in paragraph (c)(2) of this section.

 $A_{\rm t} = {
m venturi~throat~cross\text{-sectional~area}}.$ 

 $p_{\rm in} = {
m venturi~inlet~absolute~static~pressure}.$ 

$$\begin{split} Z &= \text{compressibility factor.} \\ M_{\text{mix}} &= \text{molar mass of gas mixture.} \end{split}$$

R = molar gas constant.

 $T_{\rm in}$  = venturi inlet absolute temperature.

(1) Using the data collected in §1065.340, calculate  $C_d$  using the following equation:

$$C_{\rm d} = \dot{n}_{\rm ref} \cdot \frac{\sqrt{Z \cdot M_{\rm mix} \cdot R \cdot T_{\rm in}}}{C_{\rm f} \cdot A_{\rm t} \cdot p_{\rm in}}$$

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Where:

 $\dot{n}_{\rm ref}$  = a reference molar flow rate.

(2) Determine  $C_{\rm f}$  using one of the following methods:

(i) For CFV flow meters only, determine  $C_{\rm fCFV}$  from the following table based on your values for  $\beta$  and  $\gamma$ , using linear interpolation to find intermediate values:

Table 2 of § 1065.640— $C_{\rm fCFV}$  Versus  $\beta$  and  $\gamma$  for CFV Flow Meters

| $C_{ m fCFV}$ |                             |  |  |
|---------------|-----------------------------|--|--|
| β             | γ <sub>exh</sub> =<br>1.385 | $\gamma_{\text{dexh}} = $ $\gamma_{\text{air}} = $ 1.399 |  |
| 0.000         | 0.6822                      | 0.6846   |  |
| 0.400         | 0.6857                      | 0.6881   |  |
| 0.500         | 0.6910                      | 0.6934   |  |
| 0.550         | 0.6953                      | 0.6977   |  |
| 0.600         | 0.7011                      | 0.7036   |  |
| 0.625         | 0.7047                      | 0.7072   |  |
| 0.650         | 0.7089                      | 0.7114   |  |

Table 2 of § 1065.640— $C_{\rm fCFV}$  Versus  $\beta$  and  $\gamma$  For CFV Flow Meters—Continued

| $\mathcal{C}_{	ext{fCFV}}$ |                          |  |  |
|----------------------------|--------------------------|--|--|
| β                          | γ <sub>exh</sub> = 1.385 | $\gamma_{\text{dexh}} = $ $\gamma_{\text{air}} = $ 1.399 |  |
| 0.675                      | 0.7137                   | 0.7163   |  |
| 0.700                      | 0.7193                   | 0.7219   |  |
| 0.720                      | 0.7245                   | 0.7271   |  |
| 0.740                      | 0.7303                   | 0.7329   |  |
| 0.760                      | 0.7368                   | 0.7395   |  |
| 0.770                      | 0.7404                   | 0.7431   |  |
| 0.780                      | 0.7442                   | 0.7470   |  |
| 0.790                      | 0.7483                   | 0.7511   |  |
| 0.800                      | 0.7527                   | 0.7555   |  |
| 0.810                      | 0.7573                   | 0.7602   |  |
| 0.820                      | 0.7624                   | 0.7652   |  |
| 0.830                      | 0.7677                   | 0.7707   |  |
| 0.840                      | 0.7735                   | 0.7765   |  |
| 0.850                      | 0.7798                   | 0.7828   |  |

(ii) For any CFV or SSV flow meter, you may use the following equation to calculate  $C_{\mathbf{f}}$ :

$$C_{\rm f} = \left[\frac{2 \cdot \gamma \cdot \left(r^{\frac{\gamma-1}{\gamma}} - 1\right)}{\left(\gamma - 1\right) \cdot \left(\beta^4 - r^{\frac{-2}{\gamma}}\right)}\right]^{\frac{1}{2}}$$

Eq. 1065.640-6

Where:

- $\gamma$  = isentropic exponent. For an ideal gas, this is the ratio of specific heats of the gas mixture.  $C_0/C_{11}$ .
- gas mixture,  $C_P/C_v$ . r = pressure ratio, as determined in paragraph (c)(3) of this section.
- $\beta$  = ratio of venturi throat to inlet diameters.
  - (3) Calculate r as follows:
- (i) For SSV systems only, calculate  $r_{\rm SSV}$  using the following equation:

$$r_{\rm SSV} = 1 - \frac{\Delta p_{\rm SSV}}{p_{\rm in}}$$

### Eq. 1065.640-7

Where:

 $\Delta p_{\rm SSV}$  = Differential static pressure; venturi inlet minus venturi throat.

(ii) For CFV systems only, calculate  $r_{\rm CFV}$  iteratively using the following equation:

$$r_{\text{CFV}}^{\frac{1-\gamma}{\gamma}} + \left(\frac{\gamma-1}{2}\right) \cdot \beta^4 \cdot r_{\text{CFV}}^{\frac{2}{\gamma}} = \frac{\gamma+1}{2}$$

### Eq. 1065.640-8

- (4) You may make any of the following simplifying assumptions of the governing equations, or you may use good engineering judgment to develop more appropriate values for your testing:
- (i) For emission testing over the full ranges of raw exhaust, diluted exhaust and dilution air, you may assume that the gas mixture behaves as an ideal gas: Z = 1.
- (ii) For the full range of raw exhaust you may assume a constant ratio of specific heats of  $\gamma = 1.385$ .
- (iii) For the full range of diluted exhaust and air (e.g., calibration air or dilution air), you may assume a constant ratio of specific heats of  $\gamma = 1.399$ .
- (iv) For the full range of diluted exhaust and air, you may assume the molar mass of the mixture is a function only of the amount of water in the dilution air or calibration air,  $x_{\rm H2O}$ , determined as described in §1065.645, as follows:

$$M_{\text{mix}} = M_{\text{air}} \cdot (1 - x_{\text{H2O}}) + M_{\text{H2O}} \cdot x_{\text{H2O}}$$

### Eq. 1065.640-9

Example:

 $M_{\rm air} = 28.96559 \ {
m g/mol}$  $x_{
m H2O} = 0.0169 \ {
m mol/mol}$ 

 $M_{\rm H2O} = 18.01528 \text{ g/mol}$ 

 $M_{\rm mix} \,=\, 28.96559\,\cdot\,(1\,-\,0.0169)\,+\,18.01528\,\cdot\,0.0169$ 

 $M_{\rm mix}=28.7805~{\rm g/mol}$ 

(v) For the full range of diluted exhaust and air, you may assume a constant molar mass of the mixture,  $M_{\text{mix}}$ , for all calibration and all testing as long as your assumed molar mass differs no more than ±1% from the estimated minimum and maximum molar mass during calibration and testing. You may assume this, using good engineering judgment, if you sufficiently control the amount of water in calibration air and in dilution air or if you remove sufficient water from both calibration air and dilution air. The following table gives examples of permissible ranges of dilution air dewpoint versus calibration air dewpoint:

Table 3 of § 1065.640—Examples of Dilution Air and Calibration Air Dewpoints at Which You May Assume a Constant  $M_{\rm mix}$ 

| If calibration $T_{ m dew}$ ( °C) is | assume the following constant $M_{\rm mix}$ (g/mol) | for the following ranges of $T_{\rm dew}$ ( °C) during emission tests <sup>a</sup> |
|--------------------------------------|---|--|
| dry                                  | 28.96559<br>28.89263<br>28.86148<br>28.81911        | dry to 18.<br>dry to 21.<br>dry to 22.<br>dry to 24.                               |
| 15                                   | 28.76224<br>28.68685<br>28.58806<br>28.46005        | dry to 26.<br>- 8 to 28.<br>12 to 31.<br>23 to 34.                                 |

<sup>a</sup> Range valid for all calibration and emission testing over the atmospheric pressure range (80.000 to 103.325) kPa.

(5) The following example illustrates the use of the governing equations to calculate the discharge coefficient,  $C_{\rm d}$  of an SSV flow meter at one reference flow meter value. Note that calculating  $C_{\rm d}$  for a CFV flow meter would be similar, except that  $C_{\rm f}$  would be determined from Table 2 of this section or calculated iteratively using values of  $\beta$ 

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and  $\gamma$  as described in paragraph (c)(2) of  $R = 8.314472 \text{ J/(mol \cdot K)} = 8.314472 \text{ (m}^2 \cdot \text{kg)/(s}^2 \text{ s}^2 \text{ m}^2 \text{ J/(mol \cdot K)} = 8.314472 \text{ J$ this section.

Example:

 $\dot{n}_{\rm ref}$  = 57.625 mol/s

Z = 1

 $M_{\text{mix}} = 28.7805 \text{ g/mol} = 0.0287805 \text{ kg/mol}$ 

 $\cdot \text{ mol} \cdot \text{K})$   $T_{\text{in}} = 298.15 \text{ K}$ 

 $A_{\rm t} = 0.01824 \text{ m}^2$ 

 $p_{\rm in} = 99.132~{\rm kPa} = 99132.0~{\rm Pa} = 99132~{\rm kg/(m\,\cdot\,s^2)}$ 

 $\gamma = 1.399$  $\beta = 0.8$ 

 $\Delta p = 2.312 \text{ kPa}$ 

$$R = 8.314472 \text{ J/(mol \cdot K)} = 8.314472 \text{ (m}^2 \cdot \text{kg)/(s}^2 \cdot \text{mol \cdot K)}$$

$$T_{\rm in} = 298.15 \ {\rm K}$$

$$A_t = 0.01824 \text{ m}^2$$

$$p_{\rm in} = 99.132 \text{ kPa} = 99132.0 \text{ Pa} = 99132 \text{ kg/(m·s}^2)$$

$$\gamma = 1.399$$

$$\beta = 0.8$$

$$\Delta p = 2.312 \text{ kPa}$$

$$r_{\rm SSV} = 1 - \frac{2.312}{99.132} = 0.977$$

$$C_{\rm f} = \left[ \frac{2 \cdot 1.399 \cdot \left( 0.977^{\frac{1.399 - 1}{1.399}} - 1 \right)}{\left( 1.399 - 1 \right) \cdot \left( 0.8^4 - 0.977^{\frac{-2}{1.399}} \right)} \right]^{\frac{1}{2}}$$

$$C_{\rm f} = 0.274$$

$$C_{\rm d} = 57.625 \cdot \frac{\sqrt{1 \cdot 0.0287805 \cdot 8.314472 \cdot 298.15}}{0.274 \cdot 0.01824 \cdot 99132.0}$$

$$C_{\rm d} = 0.982$$

- (d) SSV calibration. Perform the following steps to calibrate an SSV flow meter:
- (1) Calculate the Reynolds number, Re#, for each reference molar flow rate,  $\dot{n}_{\rm ref}$ , using the throat diameter of the

venturi,  $d_t$ . Because the dynamic viscosity,  $\mu$ , is needed to compute  $Re^{\mu}$ , you may use your own fluid viscosity model to determine  $\mu$  for your calibration gas (usually air), using good engineering

judgment. Alternatively, you may use the Sutherland three-coefficient viscosity model to approximate  $\mu$ , as shown in the following sample calculation for  $Re^{\#}$ :

$$Re^{\#} = \frac{4 \cdot M_{\text{mix}} \cdot \dot{n}_{\text{ref}}}{\pi \cdot d_{\text{t}} \cdot \mu}$$

Eq. 1065.640-10

Where, using the Sutherland three-coefficient viscosity model:

$$\mu = \mu_0 \cdot \left(\frac{T_{\text{in}}}{T_0}\right)^{\frac{3}{2}} \cdot \left(\frac{T_0 + S}{T_{\text{in}} + S}\right)$$

Eq. 1065.640-11

Where:

 $T_0$  = Sutherland reference temperature.

 $\mu_0$  = Sutherland reference viscosity.

S =Sutherland constant.

TABLE 4 OF § 1065.640—SUTHERLAND THREE-COEFFICIENT VISCOSITY MODEL PARAMETERS

| Gas <sup>a</sup> | $\mu_0$   | To                              | s                                | Temp range within ±2% error b   | Pressure<br>limit <sup>b</sup>             |
|------------------|---|---------------------------------|----------------------------------|---|--|
|                  | kg/(m · s)  | К                               | к                                | К   | kPa  |
| Air              | 1.716 · 10 <sup>-5</sup><br>1.370 · 10 <sup>-5</sup><br>1.12 · 10 <sup>-5</sup><br>1.919 · 10 <sup>-5</sup><br>1.663 · 10 <sup>-5</sup> | 273<br>273<br>350<br>273<br>273 | 111<br>222<br>1064<br>139<br>107 | 170 to 1900<br>190 to 1700<br>360 to 1500<br>190 to 2000<br>100 to 1500 | ≤1800<br>≤3600<br>≤10000<br>≤2500<br>≤1600 |

<sup>&</sup>lt;sup>a</sup> Use tabulated parameters only for the pure gases, as listed. Do not combine parameters in calculations to calculate viscosities of gas mixtures.
<sup>b</sup> The model results are valid only for ambient conditions in the specified ranges.

Example:

 $\mu_0 = 1.716 \cdot 10^{-5} \text{ kg/(m} \cdot \text{s)}$ 

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$$T_0 = 273 \text{ K}$$

$$S = 111 \text{ K}$$

$$\mu = 1.716 \cdot 10^{-5} \cdot \left(\frac{298.15}{273}\right)^{\frac{3}{2}} \cdot \left(\frac{273 + 111}{298.15 + 111}\right)$$

$$\mu = 1.838 \cdot 10^{-5} \text{ kg/(m·s)}$$

$$M_{\rm mix} = 28.7805 \text{ g/mol}$$

$$\dot{n}_{\rm ref} = 57.625 \text{ mol/s}$$

$$d_{\rm t} = 152.4 \text{ mm} = 0.1524 \text{ m}$$

$$T_{\rm in} = 298.15 \ {\rm K}$$

$$Re^{\#} = \frac{4 \cdot 28.7805 \cdot 57.625}{3.14159 \cdot 0.1524 \cdot 1.838 \cdot 10^{-5}}$$

$$Re^{\#} = 7.538 \cdot 10^{8}$$

(2) Create an equation for  $C_d$  as a function of  $Re^{\#}$ , using paired values of the two quantities. The equation may involve any mathematical expression, including a polynomial or a power series. The following equation is an example of a commonly used mathematical expression for relating  $C_d$  and  $Re^{\#}$ :

$$C_{\rm d} = a_0 - a_1 \cdot \sqrt{\frac{10^6}{Re^{\#}}}$$

Eq. 1065.640-12

- (3) Perform a least-squares regression analysis to determine the best-fit coefficients for the equation and calculate the equation's regression statistic, *SEE*, according to §1065.602.
- (4) If the equation meets the criterion of  $SEE \leq 0.5\% \cdot C_{dmax}$ , you may use the equation to determine  $C_{d}$  for

emission tests, as described in §1065.642.

(5) If the equation does not meet the specified statistical criterion, you may use good engineering judgment to omit calibration data points; however you must use at least seven calibration data points to demonstrate that you

meet the statistical criterion. This will usually involve narrowing the range of flow rates for a better curve fit.

- (6) Take corrective action if the equation does not meet the specified statistical criterion even after omitting calibration data points. For example, select another mathematical expression for the  $C_d$  versus  $Re^{\#}$  equation, check for leaks, or repeat the calibration process. If you must repeat the process, we recommend applying tighter tolerances to measurements and allowing more time for flows to stabilize.
- (7) Once you have an equation that meets the specified statistical criterion, you may use the equation only for the corresponding range of  $Re^{\#}$ .
- (e) CFV calibration. Some CFV flow meters consist of a single venturi and some consist of multiple venturis, where different combinations of venturis are used to meter different flow rates. For CFV flow meters that consist of multiple venturis, either calibrate each venturi independently to determine a separate discharge coefficient,  $C_{\rm d}$ , for each venturi, or cali-

brate each combination of venturis as one venturi. In the case where you calibrate a combination of venturis, use the sum of the active venturi throat areas as  $A_t$ , the square root of the sum of the squares of the active venturi throat diameters as  $d_t$ , and the ratio of the venturi throat to inlet diameters as the ratio of the square root of the sum of the active venturi throat diameters  $(d_t)$  to the diameter of the common entrance to all the venturis. (D). To determine the  $C_d$  for a single venturi or a single combination of venturis, perform the following steps:

- (1) Use the data collected at each calibration set point to calculate an individual  $C_{\rm d}$  for each point using Eq. 1065.640-4.
- (2) Calculate the mean and standard deviation of all the  $C_{\rm d}$  values according to Eqs. 1065.602–1 and 1065.602–2.
- (3) If the standard deviation of all the  $C_{\rm d}$  values is less than or equal to 0.3% of the mean  $C_{\rm d}$ , use the mean  $C_{\rm d}$  in Eq. 1065.642–4, and use the CFV only up to the highest r measured during calibration using the following equation:

$$r = 1 - \frac{\Delta p_{\text{CFV}}}{p_{\text{in}}}$$

## Eq. 1065.640-13

Where:

 $\Delta p_{\mathrm{CFV}}$  = Differential static pressure; venturi inlet minus venturi outlet.

- (4) If the standard deviation of all the  $C_{\rm d}$  values exceeds 0.3% of the mean  $C_{\rm d}$ , omit the  $C_{\rm d}$  value corresponding to the data point collected at the highest r measured during calibration.
- (5) If the number of remaining data points is less than seven, take corrective action by checking your calibration data or repeating the calibration process. If you repeat the calibration process, we recommend checking for leaks, applying tighter tolerances to measurements and allowing more time for flows to stabilize.
- (6) If the number of remaining  $C_d$  values is seven or greater, recalculate the

mean and standard deviation of the remaining  $C_d$  values.

- (7) If the standard deviation of the remaining  $C_{\rm d}$  values is less than or equal to 0.3% of the mean of the remaining  $C_{\rm d}$ , use that mean  $C_{\rm d}$  in Eq. 1065.642-4, and use the CFV values only up to the highest r associated with the remaining  $C_{\rm d}$ .
- (8) If the standard deviation of the remaining  $C_{\rm d}$  still exceeds 0.3% of the mean of the remaining  $C_{\rm d}$  values, repeat the steps in paragraph (e)(4) through (8) of this section.

[79 FR 23785, Apr. 28, 2014]